## Modelling Pt<sup>II</sup>(Ligand) – Amyloid-β Interactions: Prediction of Ligand Effects

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Alzheimer's disease (AD) is a neurodegenerative condition associated with progressive cognitive decline in patients. The causes and development of AD are poorly understood, but one hallmark of AD is the presence of amyloid plaques. The early stages of the A $\beta$  aggregation process have therefore become a target for the development of AD-therapeutics [1]. The N-terminal domain of A $\beta$  contains His-rich high-affinity metal binding sites, responsible for physiological coordination of Cu<sup>II</sup>/Zn. One route to AD-therapeutics involves disrupting coordination by using compounds that occupy these binding sites, hindering the A $\beta$  aggregation process. Recently, Barnham *et al.* showed that Pt<sup>II</sup>(phenanthroline) complexes inhibit A $\beta$  aggregation and limit its neurotoxicity *in vitro* [2].

In this work, Ligand Field Molecular Mechanics (LFMM), DFT and semi empirical methods are applied to a series of  $Pt^{II}$ -Ligand systems binding to the N-terminal domain of the A $\beta$  peptide. Molecular dynamics (LFMM/AMBER) is used to explore the conformational freedom of the peptide fragment, and identifies favourable  $Pt^{II}$ -binding modes and peptide conformations.  $Pt^{II}$  coordination depends on the nature of the ligand, providing evidence that binding mode may be controlled by ligand design. Structural analysis of the sampled  $Pt^{II}$ -A $\beta$  conformations shows that platinum coordination disrupts existing secondary structure in A $\beta$  and promotes formation of ligand-specific turn-type secondary structure. [3][4]

References:

- [1] D. Valensin, C. Gabbiani, L. Messori, Coord. Chem. Rev., 256, 2357, (2012).
- [2] K.J. Barnham, V.B. Kenche, G.D. Ciccotosto, D.P. Smith, D.J. Tew, X. Lu, K.Perez, G.A. Cranston, T.J. Johanssen, I. Volitakis, A.I. Bush, C.L. Masters, A.R. White, J.P. Smith, R.A. Cherny, R. Cappai, *Proc. Natl. Acad. Sci. U.S.A.*, **105**, 6813, (2008).
- [3] M. Turner, J.A. Platts and R.J. Deeth, J. Chem. Theory Comput., 12, 1385, (2016).
- [4] M. Turner, R.J. Deeth and J.A. Platts, J. Inorg. Biochem., 173, 44, (2017).